Chemistry of Life Unit: Key Idea You are what you eat ... or have eaten

Quaker Quick Oatmeal

Nutritic Serving Size 1/2 cup dry	/ (40g)	acts
Servings Per Container	about 79	
Amount Per Serving		
	Cereal Alone	with 1/2 cup Vitamin A&D Fortified Skim Milk
Calories	150	190
Calories from Fat	25	25
		% Daily Value**
Total Fat 3g*	5%	5%
Saturated Fat 0.5g	2%	2%
Polyunsaturated Fat	1g	
Monounsaturated Fat 1g		
Cholesterol 0mg	0%	0%
Sodium 0mg	0%	3%
Total Carbohydrate 27	'g 9%	11%
Dietary Fiber 4g	15%	15%
Soluble Fiber 2g		
Insoluble Fiber 2g		
Suigars 1g		
Protein 5g		

Grape Nuts

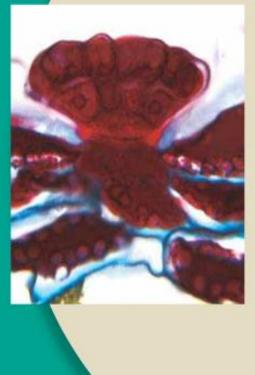
Nutrition Facts
Serving Size 1/2 cup (58g)
Servings Per Container about 12

Amount	Gene	al with	
Per Serving	Carrel St.	in Milk	
Calories	200	240	
Calories from Fat	10	10	
	% Daily V	alue""	
Total Fat 1g*	2%	2%	
Saturated Fat 0g	0%	0%	
Cholesteral Omg	0%	0%	
Sodium 350mg	15%	17%	
Potassium 160mg	5%	10%	
Total Carbohydra e47	7g 16%	18%	
Dietary Fiber 5g	21%	21%	
Sugars 7g			
Other Carbohydrate 35g			
Protein 6g			



Connecting CONCEPTS

Like other carnivores, the Venus flytrap eats animals to get nutrients that it needs to make molecules such as proteins and nucleic acids. Other chemical compounds made by the plant's cells enable the Venus flytrap to digest the animals that it eats. These chemicals are similar to the chemicals that allow you to digest the food that you eat.



Cell Function The Venus flytrap has specialized cells on the surfaces of its leaves. Some of these cells allow the plant to snap shut on its prey within 0.5 seconds. Other cells, such as those that appear purple in this light micrograph, secrete digestive chemicals that allow the plant to consume its prey. (LM; magnification 500×)

Chapter 2: Chemistry of Life

Prefix Preview:

Identify the Latin "root word" that matches each prefix

Prefix	Latin "Root Word"
ONE	Mono
TWO	Di
THREE	Tri
MANY	Poly

All living things contain the same 4 Carbon-based molecules:

C video

- The Biomolecules are called POLYMERS.
 - All Polymers are made up of monomers, or "building blocks."
 - They are all <u>built</u> with the same process: Dehydration Synthesis (Condensation)

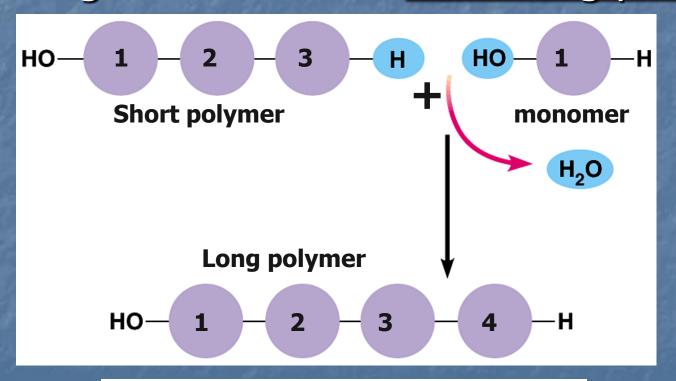
$$BB + BB = polymer + H_2O$$

All are <u>broken down</u> by the same process: <u>HYDROLYSIS</u>:

polymer +
$$H_2O$$
 = BB + BB

A closer look at the chemistry

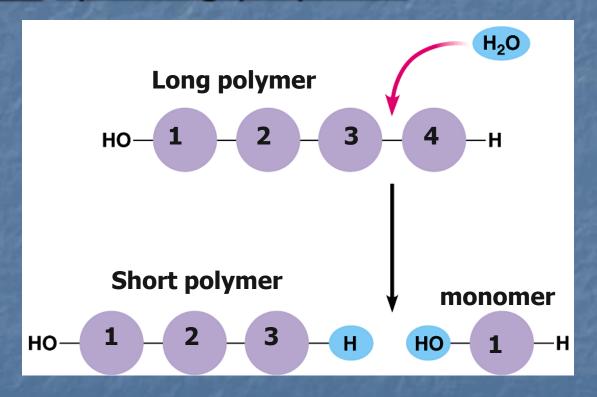
Condensation = <u>WATER</u> is formed during a reaction that <u>builds</u> big polymers



BB + BB → Polymer + Water

A closer look at the chemistry

Hydrolysis = Add <u>WATER</u> and then split apart big polymers



Polymer + Water → BB + BB

1. You drop your cell phone and it breaks into 15 pieces?

2. You staple 8 pages of paper together?

3. You build a castle out of legos?

4. You use a knife to slice a carrot into many pieces?

5. You drink some water and then digest a donut in your stomach?

6. Your fingernails grow longer in length as your nail bed cells connect protein monomers to build protein polymers?

1. You drop your cell phone and it breaks into 15 pieces?

Hydrolysis

- 2. You staple 8 pages of paper together?

 Condensation
- 3. You build a castle out of legos? **Condensation**
- 4. You use a knife to slice a carrot into many pieces? Hydrolysis

5. You drink some water and then digest a donut in your stomach?

Hydrolysis

6. Your fingernails grow longer in length as your nail bed cells connect protein monomers to build protein polymers?

Condensation

VISUAL VOCAB

Each smaller molecule is a subunit called a monomer.

mono- = one poly- = many

monomer

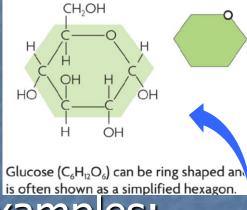
polymer

A **polymer** is a molecule that contains many monomers bonded together.

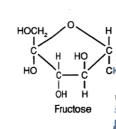
1) Carbohydrates

BUILDING BLOCK (one sugar subunit or monomer): Monosaccharide

Shape = Hexagon ring



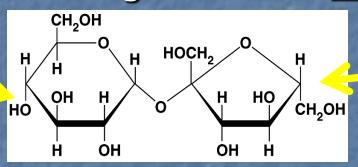
- 2 common monosaccharide examples:
- 1) All carbohydrates in the diet are digested until they are absorbed into the blood as <u>Glucose</u>
- 2) The simple sugar found in fruit is Fructose



1) Carbohydrates

- Disaccharide- Two linked sugar molecules
- 2 common disaccharide examples:
- Table sugar is called <u>Sucrose</u>

Glucose

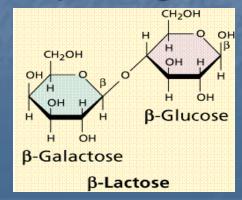


Fructose



2. The simple sugar in milk is called

Lactose





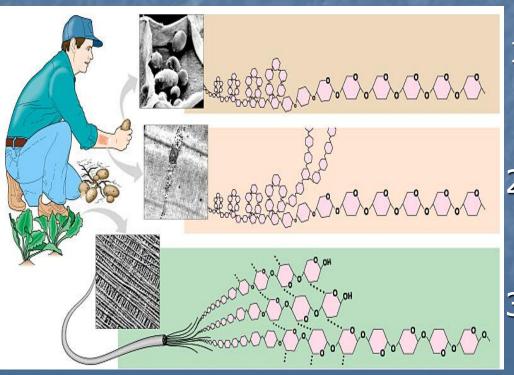
Disaccharides V

1) Carbohydrates

Polysaccharide - many linked sugars

3 common polysaccharide examples

Poly S video



1) In potatoes = Starch

2) In muscle = Glycogen

3) In plant cell walls = Cellulose

Important things to know about Carbs:

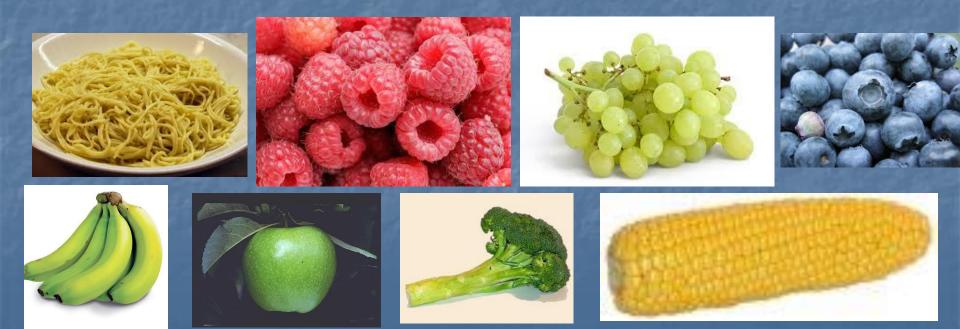
- Many sugars usually end in "OSE "
 - Glucose, sucrose, lactose, fructose, cellulose
- Polysaccharides have various <u>functions</u>:
 - <u>Starch</u> plant long-term Energy storage
 - Glycogen human long-term Energy storage in liver and muscles
 - Cellulose plant cell walls (cotton fibers) (indigestible by humans = dietary <u>fiber</u>)
 - chitin protective animal coverings (crayfish, shrimp, beetles)



Important things to know about Carbs:

- Easy to use form of stored energy for cells
- What do all the carbohydrate-containing foods below have in common?

Made by plants that captured solar energy



What do these carbs have in common?









- Color = white
- Glycemic Index = HIGH
- Health = BAD WHY?

Blood glucose spikes → Blood insufin spikes →

Blood glucose plummets Triggers **HUNGER** >

Overeating → Obesity →

- ↑ 3 Major Health Risks:
- 1) Heart Disease (Insulin 1) artery inflammation and plaque buildup)
- 2) Type 2 Diabetes (↑↓↑↓ Insulin → Insulin resistance & ↑blood glucose levels)

3) Cancer (Insulin triggers other hormones that <a href="https://cell.doi.org/10.2016/j.cell.

Bottom line: See WHITE carbs: think FAT

See the carbohydrate in this Coke label?



How do they make HFCS?

- Break down corn starch into glucose monomers
- Convert most glucose into fructose

WHY?

- Fructose is sweeter than glucose
- Fructose is easier to work with
- Fructose is cheaper than sucrose (table sugar)

Why does HFCS concern nutrition experts?

- Fructose blocks the burning of <u>Body fat</u>
- Fructose stimulates production of liver <u>fat</u>
 which → ↑Insulin <u>resistance</u>
- Fructose stimulates <u>appetite</u>



WOW ... cutting 1 16-ounce soda each day = losing 20 lbs in 1 year?

"Superstar" Carbohydrate foods include:

beans , whole grains , Fruits & Vegetables

These foods are full of natural chemicals which help us by:

- 1) . inflammation
- 2) Neutralize harmful "free radical" molecules that constantly harm cells
- 3) . 1 immune system
- 4) . VBody toxins

Let's Review

C video

Cn H video

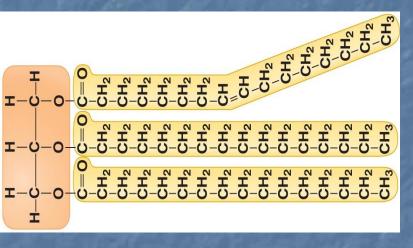
Disaccharides V

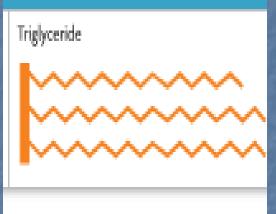
Poly S video

Lipid video

2) Lipids

BUILDING BLOCK: 1 Glycerol head with three fatty acid tails







Lipid video

Important things to know about Lipids:

NOT soluble in water (<u>nonpolar</u>)











- 3 main types of Lipids
- 1) <u>fats</u>
- 2) <u>oils</u>
- 3) <u>waxes</u>









Important things to know about Lipids:

Foods high in Lipids: Fried foods

Butter



Whole cheese







Red meat

Lipids have various **Functions** in cells:



Fats – long-term Energy storage



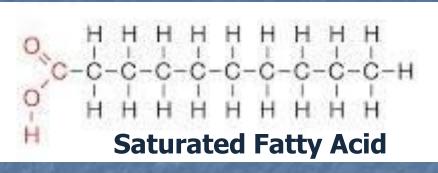


 Steroids – <u>cholesterol and hormones (estrogen,</u> <u>testosterone)</u>



<u>Pigments – light-absorbing chlorophyll in plants</u>

Notice any <u>difference</u> between these 2 Fatty Acids?



Which fatty acid is full of hydrogen?

- Saturated fat- Solid at room temperature (like butter, lard, solid fat in red meat)
 - Health = Bad → linked to cholesterol buildup in arteries

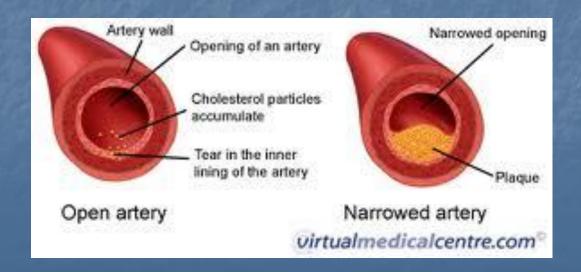


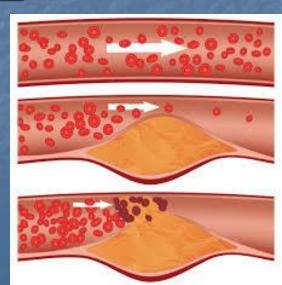






- Here's how the "4-legged "Saturated fats cause cardiovascular trouble:
- 1) Stimulate production of the <u>LDL</u> "bad" cholesterol
- 2) Cripple the HDL "good" cholesterol leading to artery wall inflammation
- 3) Block appetite-suppressing brain hormones which leads to 3 days of <a href="https://overeating.com/overeatin





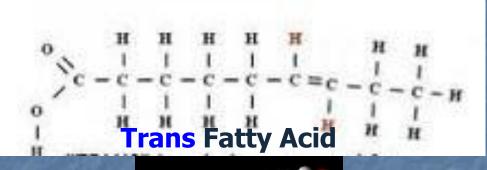
Notice any <u>difference</u> between these 2 Fatty Acids?

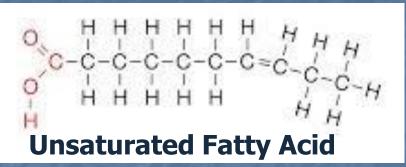
Which fatty acid has room to bond with more hydrogen atoms?

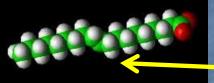
Unsaturated fat-liquid at room temperature (like plant oils)

 \rightarrow Health \bigcirc K \rightarrow less risk for cholesterol buildup

Notice any <u>difference</u> between these 2 Fatty Acids?







That little difference in structure creates a "kink" in the trans fatty acid that causes a lot of trouble

- Trans Fat found in some oils used for frying
 - Health $\frac{3a}{}$ \rightarrow linked to
 - 1) Cardiovascular disease
 - 2) Type 2 Diabetes
 - 3) **Wental function**

Foods that are likely to have trans fats:

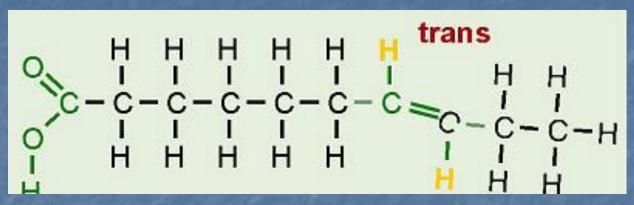
- Fried Fast foods
- Donuts, muffins
- Pie Crusts
- Crackers
- Cookies
- Chips
- Cake, cake icing
- Microwave popped corn
- Canned biscuits





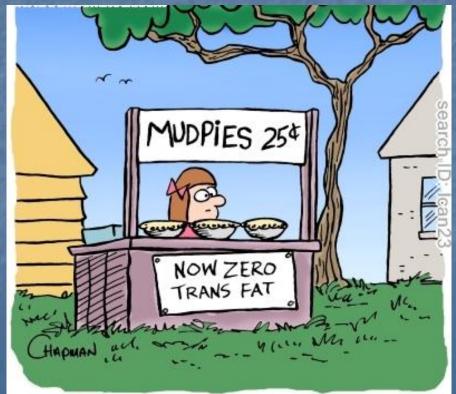








"Apparently, it's extremely bad for the heart. But I wouldn't worry – it only affects people who have one."



I always considered myself more of a butter-fly....





"Bless me father for I have sinned. I use trans fats for all of my pie crusts,"

Even McDonalds changed the "sacred" frying oil in 2008

McDonald's fries are now trans fat-free in U.S., Canada





Lipid Health Guidelines: Look for "good" fats

1) Consume monounsaturated fats













2) Get plenty of omega-3 fats to help your heart health and brain function

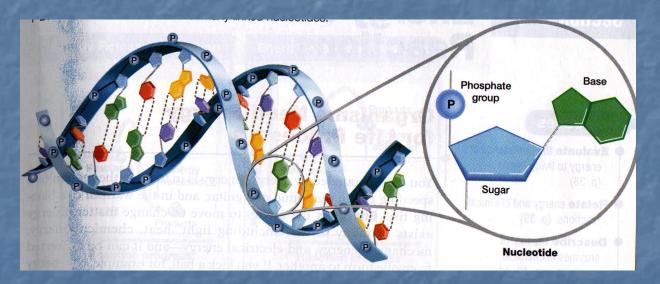






Nucleic Acids

- BUILDING BLOCK: Nucleotides = "L"
 - Phosphate
 - Sugar
 - Nitrogen base



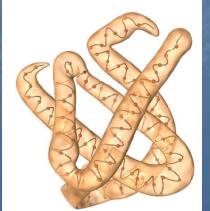
- Examples: DNA and RNA
- Function: store <u>genetic</u> information

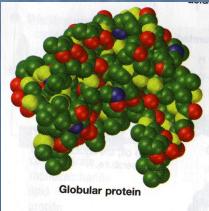
Proteins

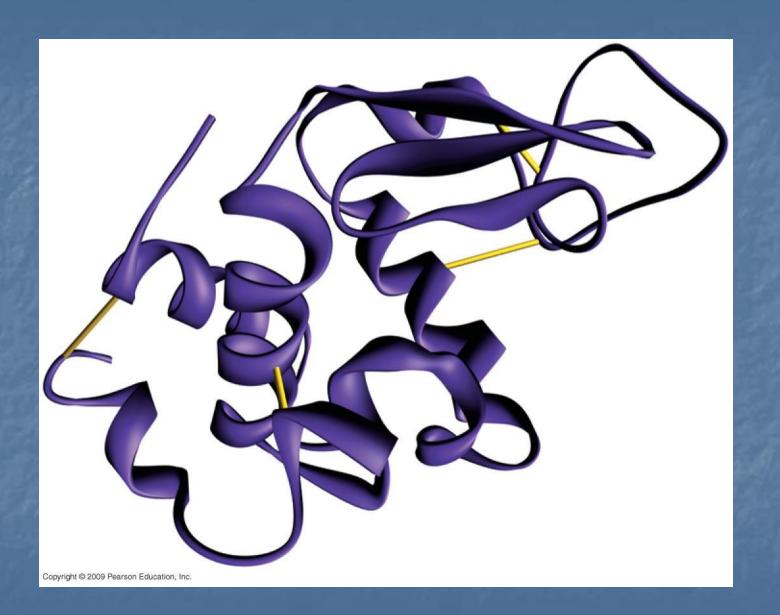
- BUILDING BLOCK: Amino Acids
 - There are 20 different types of amino acids
 - 10 are Essential → must be eaten in diet for health
 - 10 are Nonessential → can be produced by the body

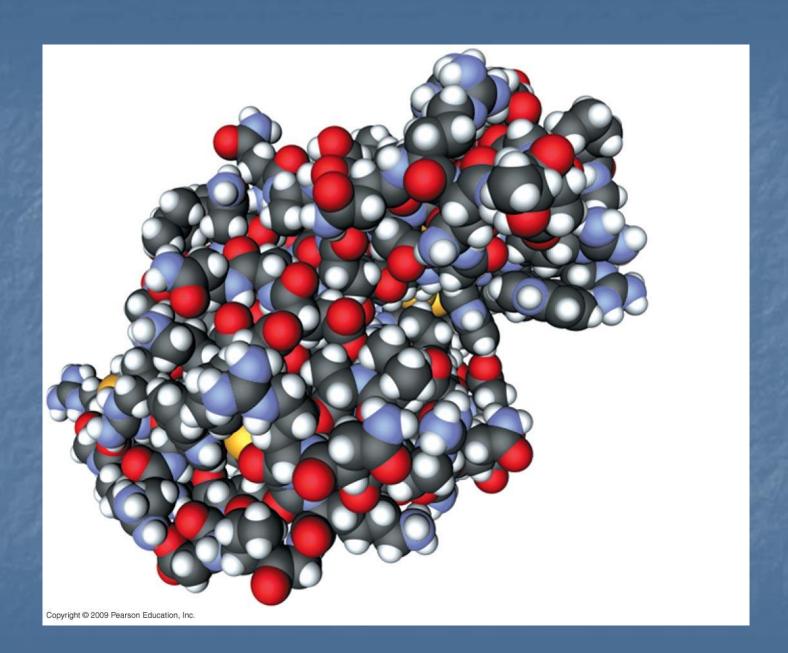


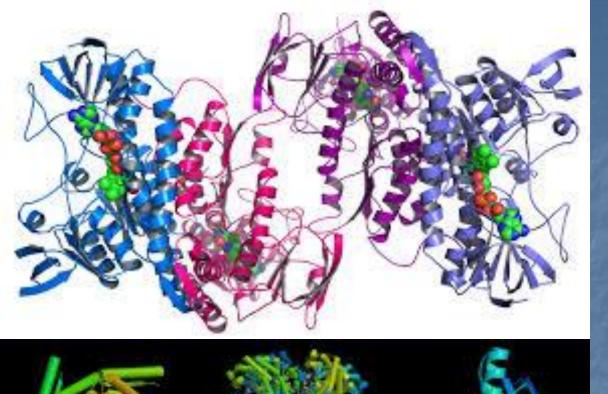
Many times, proteinsfold up into weird3D globular shapes



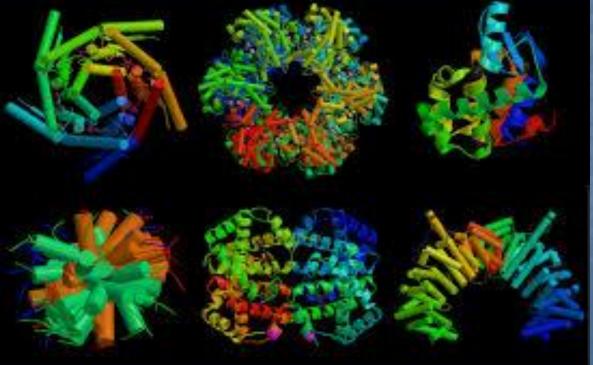








Protein intro



Protein 1

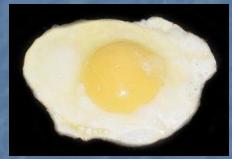
Protein 2

Protein 3

Important things to know about Proteins:

- \blacksquare AA + AA = $\boxed{\text{Pi}}$ peptide
- \blacksquare AA + AA + AA + AA= \boxed{Poly} peptide
- Sources of protein:
 - Meat, milk, eggs, legumes









Examples: (Proteins have many jobs)

<u>Collagen</u> – building material in skin, ligaments, tendons, bones

Hemoglobin

__ – carries Oxygen in the blood



Important things to know about Proteins:

- Examples: (Proteins have many jobs)
 - Insulin helps Glucose into cells
 - _____Antibodies __ fight infections and diseases
 - Various protein fibers form <u>muscle</u>tissue and help blood <u>clot</u>
 - Enzymes speed up chemical reactions in the body

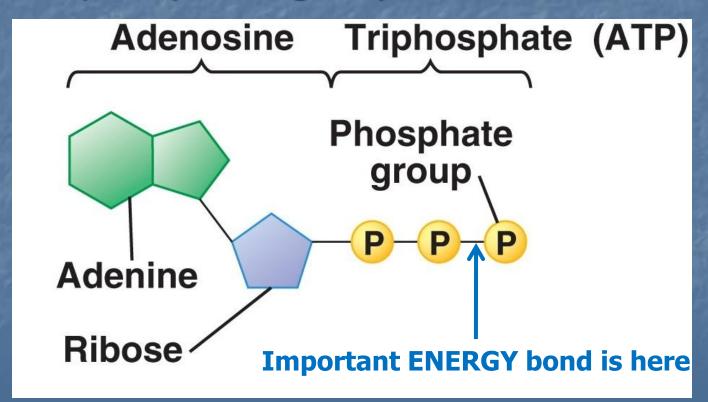
Let's Review the 4 biomolecules

Biomolecule	Monomer "BB"
Carbohydrate	Monosaccharide
Lipid	3 Fatty Acids & 1 Glycerol
Nucleic Acid	Nucleotide
Protein	Amino Acid

ATP (adenosine tri-phosphate)



 BUILDING BLOCK: a single nucleotide + 2 extra phosphate groups



Important things to know about ATP:

- Chemical Energy is stored in <u>covalent</u>
 bonds between phosphate groups
 - When the bond breaks, energy is released
- FUNCTION
 - carries <u>Energy</u> in cells
 - <u>fuels</u> most cellular activities
 (like \$1 of energy)



There are also molecules known as AMP molecules (adenosine monophosphate) and ADP molecules (adenosine diphosphate).

Remember what your *pre-*fixes correspond to when we discuss biomolecules.